

## Research Article

## A multistep preparation of 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole under microwave ir-radiations

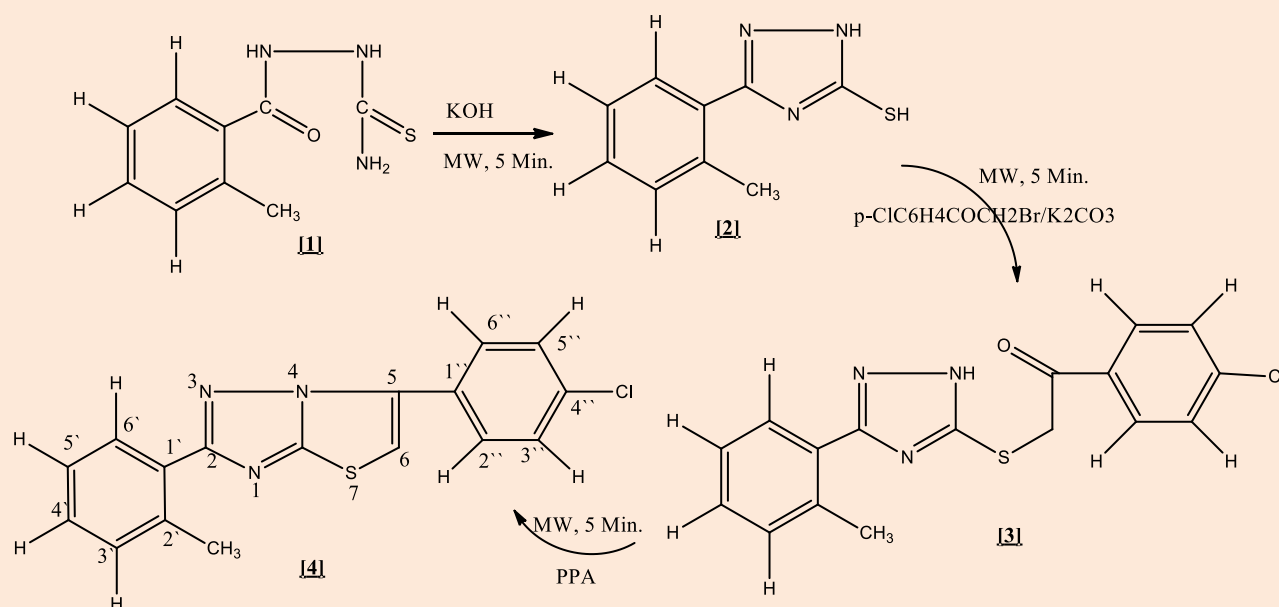
Rakesh Kumar<sup>1</sup>, Ravinder Singh<sup>2\*</sup> and Hiteshi<sup>3</sup><sup>1</sup>Department of Chemistry, Govt. College, Jhajjar<sup>2</sup>Department of Chemistry, Govt. College for Women, LakhanMajra (Rohtak)<sup>3</sup>Scholars Rosary Sen. Sec. School, Rohtak**Abstract**

The 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole having different biological activities are prepared in high yield using microwave conditions which becomes a part of green chemistry due its non pollution nature

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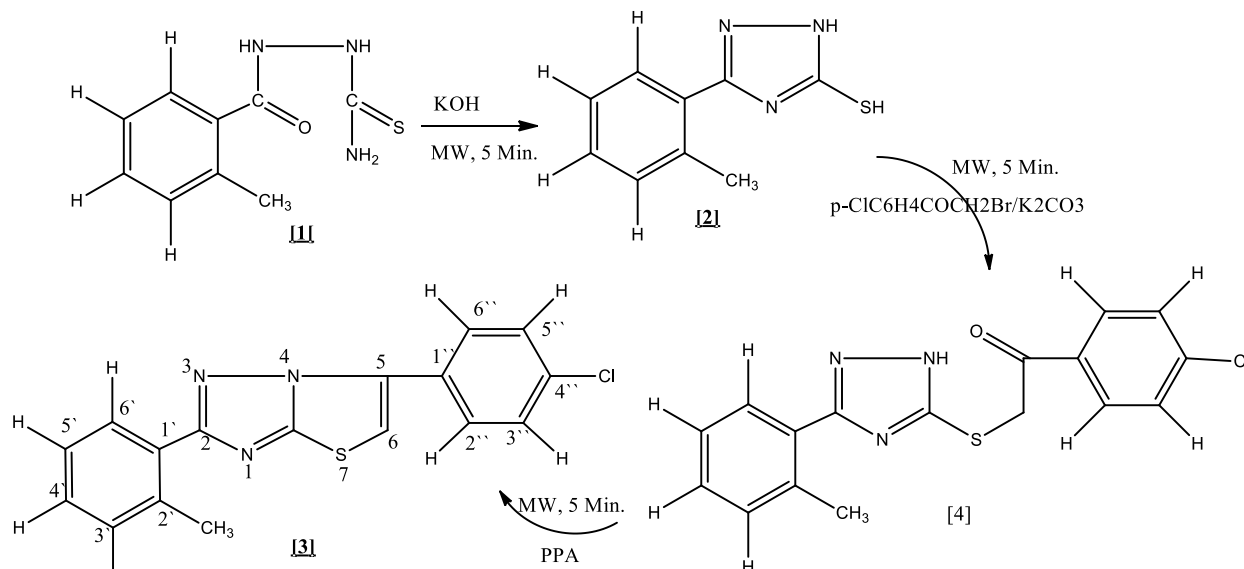
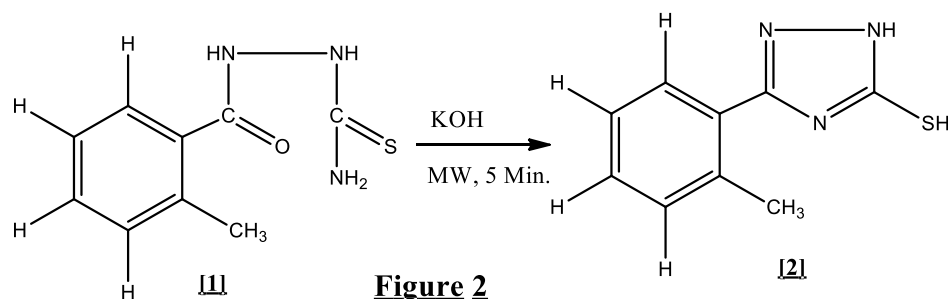
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**Keywords:** Triazole, Microwave, Aryl, Heterocyclic, Biological activity**Introduction**

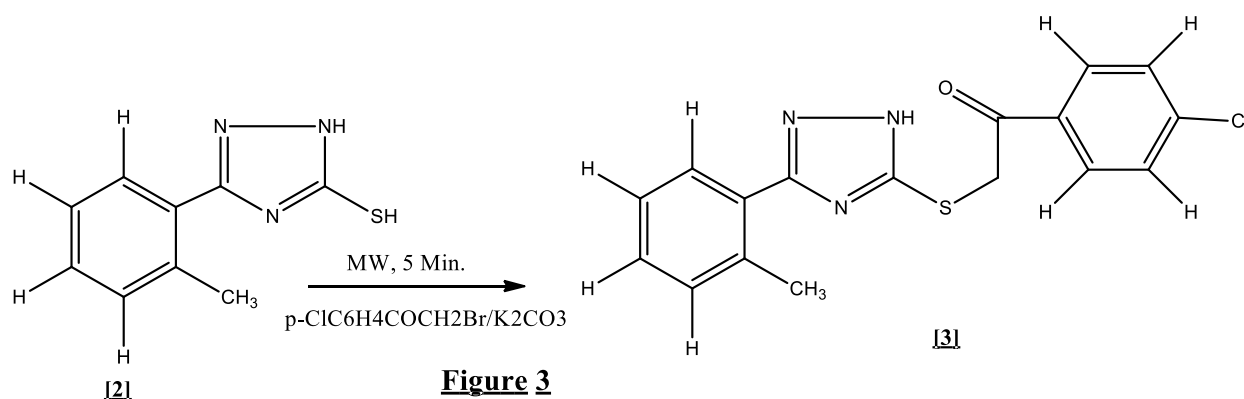
The triazoles, exhibit potent antineoplastic agent [1], bactericide and a fungicide [2], insecticidal and acaricidal activities [3]. The triazoles are previously prepared by ordinary heating using Bunsen burner which causes pollution and takes very long time for reaction completion and also have hectic workup process [4-18]. The organic reaction supported by Microwave conditions causes no pollution, reduces the reaction time, causes uniform heating of reaction material [19-28].

Our research work deals with the synthesis of 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole having different biological activities in high yield using microwave conditions which becomes a part of green chemistry due its non pollution nature (**Figure 1**).

Our research study started by reacting 4-o-methylbenzoylthiosemicarbazide[1]) with 8% potassium hydroxide solution using microwave conditions at optimum condition of 560W for 5-minutes leads to formation of 5-Mercapto-3-o-tolyl-s-triazole[2] (**Figure 2**). The 4-o-methylbenzoylthiosemicarbazide[1] was in turn obtained by reacting o-methylbenzoylhydrazide with potassium thiocyanate, conc. HCl and water using microwave conditions at optimum condition of 560W for 5-minutes.

**Figure 1****Figure 2**

Further, 5-Mercapto-3-o-tolyl-s-triazole[2] reacts with p-chlorophenacyl bromide in dry methanol using microwave conditions at optimum condition of 560W for 5-minutes to afford white compound 5-p-Chlorobenzoylmethylmercapto-3-o-tolyl-s-triazole[3] (**Figure 3**).

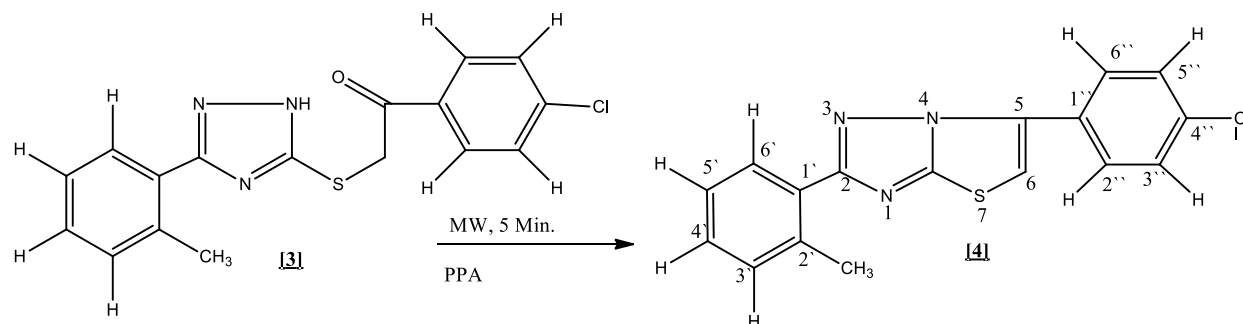
**Figure 3**

We further explored our work by reacting 5-p-Chlorobenzoylmethylmercapto-3-o-tolyl-s-triazole[3] ( $R_1 = p\text{-ClC}_6\text{H}_7$ , 1g) with PPA/ $\text{P}_2\text{O}_5$  using microwave conditions at optimum condition of 560W for 5-minutes which leads to the formation of 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole[4] (**Figure 4**).

#### Synthesis of 4-o-methylbenzoylthiosemicarbazide[1]

A mixture of o-methylbenzoylhydrazide (1.50g, 0.1 mol), potassium thiocyanate (19.4g, 0.2 mol), conc. HCl (8 ml) and water (150 ml) was irradiated using microwave conditions at optimum condition of 560W for 5-minutes. The resulting mixture was cooled and filtered to get resultant compound. The compound was further filtered, washed

using water and then crystallized using ethanol-DMF furnishing colourless shining flakes of 4-o-methylbenzoylthiosemicarbazide[1]. m.p. 185°C, yield 89% [Found : S, 15.03.C<sub>9</sub>H<sub>11</sub>N<sub>3</sub>OS requires S, 15.31%].



**Figure 4**

### Synthesis of 5-Mercapto-3-o-tolyl-s-triazole[2]

A mixture of 4-o-methylbenzoylthiosemicarbazide(I) (20.9 g, 0.1 mol) in 8% potassium hydroxide solution (300 ml) was irradiated under microwave irradiation at 560W for 5-minutes. The resulting mixture was cooled and then small amount of dil. acetic acid is added. The compound was further filtered, and then crystallized using ethanol to give white solid 5-Mercapto-3-o-tolyl-s-triazole[2]. m.p. 255°C (Lit.<sup>254</sup> m.p. 255°C), yield 77% [Found :S, 16.92.C<sub>9</sub>H<sub>9</sub>N<sub>3</sub>S requires S, 16.75%].

### Synthesis of 5-p-Chlorobenzoylmethylmercapto-3-o-tolyl-s-triazole[3]

A mixture of 5-Mercapto-3-o-tolyl-s-triazole[2] (1.91g, 0.01 mol) and p-chlorophenacyl bromide (2.34g, 0.01 mol) in dry methanol (100 ml) was treated using microwave conditions under optimum condition at 560W for 5-minutes. The resultant mixture was cooled and then small amount of aq. K<sub>2</sub>CO<sub>3</sub> is added for neutralization. The resultant compound was filtered, and crystallized using methanol to give a white solid 5-p-Chlorobenzoylmethylmercapto-3-o-tolyl-s-triazole[3]. m.p. 220°C, yield 65%. IR : 755, 830 (1, 2 & 1, 4-dsubstituted benzene rings) 1485 (C-N stretching), 1590 (C=N), 1690 (C=O), 3400 (NH) [Found: N, 12.33; S, 9.11.C<sub>17</sub>H<sub>13</sub>N<sub>3</sub>OSCl requires N, 12.26; S, 9.34%].

### Synthesis of 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole[4]

A mixture of 5-p-Chlorobenzoylmethylmercapto-3-o-tolyl-s-triazole[3] (R<sub>1</sub> = p-C<sub>1</sub>C<sub>6</sub>H<sub>7</sub>, 1g), P<sub>2</sub>O<sub>5</sub> (4.0g) and H<sub>3</sub>PO<sub>4</sub> (3 ml) was treated using microwave conditions under optimum condition at 560W for 5-minutes. The resultant mixture was cooled and then small amount of aq. K<sub>2</sub>CO<sub>3</sub> is added for neutralization. The resultant compound was filtered, and crystallized using ethanol to give colourless needles of 2-o-Tolyl-5-p-chlorophenyl thiazolo [3, 2-b]-s-triazole[4]. Yield: 72%, m.p. > 250°C. IR : 740, 830 (1, 2 & 1, 4 - disubstituted benzene), 1500 (C-N), 1580, 1600 (C = C & C =N), 3040 (C-H, Ar); H<sup>1</sup>-NMR (CDC<sub>13</sub>) : δ 2.30 (3H, s, CH<sub>3</sub>), 6.70 (1H, s, C<sub>6</sub>-H) 6.60 – 7.40 (8H, m, Ar-H). [Found :C, 62.78; H, 3.46; S, 9.68; N, 13.14.C<sub>17</sub>H<sub>12</sub>N<sub>3</sub>SCl requires C, 62.86; H, 3.39; S, 9.86; N, 12.94%].

## References

- [1] Kano, S; Noguchi T, Japan Pat., 1971, 836, 71 37; Chem. Abstr., 1972, 76, 25295g.
- [2] Hoffmann H; Hammann I, Ger. Offen., 1972, 32, 2, 173 (1972); Chem. Abstr., 1972, 76, 72525s.
- [3] Deshmukh A.A., Mody M.K., Ramalingam T. and Sattur P.B., Indian J. Chem, 1984, 23B, 793.
- [4] Rogness D. C, Larock, R.C. Tetrahedron Lett, 2009, 50, 4003.
- [5] Lin, Z; Larock, R.C., J. Org. Chem., 2006, 71, 3198.
- [6] Peddibhotla, S., Curr. Bioact. Compd., 2009, 5, 20.
- [7] Saburo K., Zasshi Y, 1972, 92(8), 935; Chem. Abstr., 1972, 77, 126492V.
- [8] Potts K.T. and Husain S., J. Org. Chem., 1971, 36, 10.
- [9] Mohan J, Indian J. Chem., 1982, 21B, 243.
- [10] Mohan J and Anjaneyulu G.S.R., Polish J. Chem., 1987, 61, 547.
- [11] Hoggarth E., J. Chem. Soc., 1952, 4811.
- [12] George T., Thilramani R. and Dabholkar D.A., Indian J. Chem., 1969, 7, 959.
- [13] Reid J.R. and Heindel N.D., J. Heterocycl. Chem., 1976, 13, 925.

- [14] Hantzsch and Weber H.J., Berdt Chem. Ges., 1987, 20, 3118.  
[15] Dodson R.M., King L.C., J. Am. Chem. Soc., 1945, 87, 2242.  
[16] King L.C. and Hlavacek R.J., J. Am. Chem. Soc., 1950, 72, 3722.  
[17] Pattanayak B.K., Rout D.N. and Mahapatra G.N., Indian J. Chem., 1978, 16B, 1030.  
[18] Chadha V.K. and Pujari H.K., Can. J. Chem., 1979, 47, 2843.  
[19] Singh R, Chemical Science Review and Letters, 4, 15(2015), 835-837.  
[20] Singh R, Rajeev Kumar, Chemical Science Review and Letters, 4, 16(2015), 937-940.  
[21] Dewan S K, Singh R, Arkivoc, 2006, Vol (ii), 41-44.  
[22] Dewan S K; Singh R; Anil Kumar, Synthetic Communications; 2004, 34(11), 2025-2029.  
[23] Dewan S K, Singh R Synthetic Communications; 2003, 33(17), 3085-3088.  
[24] Dewan S K, Singh R, Synthetic Communications; 2003, 33(17), 3081-3084.  
[25] Dewan S K, Singh R J. Indian council Chemists; 2003, 20(1), 1-3.  
[26] Dewan S K, Singh R, Indian J. Heterocyclic Chemistry; 2003, 12(1), 287-288.  
[27] Dewan S K, Singh R, Indian J. Heterocyclic Chemistry; 2002, 12(1), 173-174.  
[28] Dewan S K, Singh R, Kanwar M, oriental J. of chemistry, 2002, 18(3), 555-558.

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