# **Research Article**

# Synthesis of 6-Methoxyisatin-3-thiosemicarbanzone in presence of Mont. K-10, Mont. KSF catalysts under Microwave ir-radiations

Ravinder Singh\*

Govt. College for Women, Lakhanmajra

### Abstract

The 6-Methoxyisatin-3-thiosemicarbanzone having antihistaminic, antihyroid, antitubercular, antifungal & antibacterial activities synthesized in high yield in presence of heterogeneous Mont. K-10, Mont. KSF catalysts in shorter reaction time under microwave irradiations as compared to under simple conventional heating.

Keywords: Microwave, Aryl, heterocyclic, Conventional heating, thiosemicarbazone

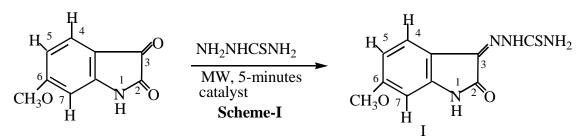
#### \*Correspondence

Author: Ravinder Singh Emails: gahlawat.ravinder@gmail.com

### Introduction

The nitrogen heterocycles exhibit antihistaminic, antithyroid, antitubercular, antifungal, antibacterial, [1-3] anthelminitics activities, antidepressants, platelet aggregation inhibitors, antineoplastic, vulcanization accelerators, photographic sensitizers [4-11] are already been synthesized by different method but they requires longer reaction time and tedious workup [12-21]. Microwave assisted reactions are gaining much more importance in synthetic organic chemistry due to dramatic reduction in time from days to hours and hours to minutes or seconds [22-23].

The present work reports the synthesis of 6-Methoxyisatin-3-thiosemicarbanzone(I) under conventional heating and in presence of heterogeneous catalysts Mont. K-10 and Mont. KSF under microwave irradiation (Scheme 1).



A mixture of 6-methoxyisatin in Anhyd. ethanol and thiosemicarbazide in a mixture of water and glacial acetic acid on conventional heating for 1 hour give 6-Methoxyisatin-3-thiosemicarbanzone(I) in 90% yield. It was found that the a mixture of 6-methoxyisatin in Anhyd. ethanol and thiosemicarbazide in a mixture of water and glacial acetic acid under microwave irradiation at 560W for 5-minutes give 6-Methoxyisatin-3-thiosemicarbanzone(I) in 95% yield.

Further it was found that the 6-methoxyisatin on reaction with thiosemicarbazide under microwave irradiation at 560W for 5-minutes in presence of heterogeneous catalysts Mont. K-10 and Mont. KSF give 6-Methoxyisatin-3-thiosemicarbanzone(I) in 97%, 98% yield respectively. It is concluded that yield of product increases under microwave heating in presence of catalyst (**Table 1**).

Sr. No.	Condition	%age yield of 6-Methoxyisatin- 3-thiosemicarbanzone	Time (in minutes)	Melting point (°C)
1	Without catalyst under conventional heating	90%	5	>250°C
2	Without catalyst under Microwave heating	95%	5	>250°C
3	With Mont.K-10 under MW heating	97%	5	>250°C
4	With Mont. KSF under MW heating	98%	5	>250°C

### **Table 1** Synthesis of 6-Methoxyisatin-3-thiosemicarbanzone under different conditions

### Experimental

All the melting points reported are uncorrected. Infrared spectra ( $v_{max}$  in cm<sup>-1</sup>) were recorded in nujol mull or KBr on a Perkin-Elmer 842/Beckman IR-20/Hitachi 215 spectrometers. The proton magnetic resonance spectra were recorded on a VXR-200 MHz or R-32 Perkin-Elmer 90 MHz spectrometer in CDC1<sub>3</sub> or DMSO-d<sub>6</sub> using tetramethylsilane (TMS) as internal reference stadnard. The chemical shifts are expressed in  $\delta$  (ppm) units downfield from TMS. Mass spectra were scanned on a Jeol JMX-DX-300 spectrometer operating at 70 eV. Carbon, hydrogen and nitrogen analyses were carried out on a Yanaco MT-3 (JAPAN) instrument. Thin layer chromatography (TLC) were performed on silica-gel plates using acetone-benzene (1:3 or 1:2) as solvent system and iodine chamber as visualizing agent.

# Typical procedure for the synthesis of 6-Methoxyisatin-3-thiosemicarbanzone(I) in presence of solvent under conventional heating

A mixture of 6-methoxyisatin (0.18g, 0.001 mol) in Anhyd. ethanol (2ml) and thiosemicarbazide (0.1g, 0.0011 mol) in a mixture of water (2 ml) and glacial acetic acid (0.5 ml) was heated for 1-hour. A yellow coloured solid formed during irradiation. The solid was filtered, washed well with water and crystallized from ethanol-DMF furnishing yellow crystals. Yield 0.234g (90%), m.p. 265°C. [Found: N, 22.68, S, 12.62.  $C_{10}H_{10}N_4O_2S$  requires N, 22.40; S, 12.80%]; IR: 825, 860 (1, 2, 4-trisubstituted benzene ring), 1115 (C=S), 1125 & 1370 (C-O-C stretching), 1620 (C=N), 1700 (C=O), 3200, 3280, 3400 (NH, NH<sub>2</sub>).

# Typical procedure for the synthesis of 6-Methoxyisatin-3-thiosemicarbanzone(I) in presence of solvent under microwave heating

A mixture of 6-methoxyisatin (0.18g, 0.001 mol) in Anhyd. ethanol (2ml) and thiosemicarbazide (0.1g, 0.0011 mol) in a mixture of water (2 ml) and glacial acetic acid (0.5 ml) was irradiated under microwave irradiation at 560W for 5-minutes. A yellow coloured solid formed during irradiation. The solid was filtered, washed well with water and crystallized from ethanol-DMF furnishing yellow crystals. Yield 0.247g (95%), m.p.  $265^{\circ}$ C. [Found: N, 22.68, S, 12.62. C<sub>10</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub>S requires N, 22.40; S, 12.80%]; IR: 825, 860 (1, 2, 4-trisubstituted benzene ring), 1115 (C=S), 1125 & 1370 (C-O-C stretching), 1620 (C=N), 1700 (C=O), 3200, 3280, 3400 (NH, NH<sub>2</sub>).

# Typical procedure for the synthesis of 6-Methoxyisatin-3-thiosemicarbanzone(I) in dry media using Mont. K-10, Mont. KSF catalyst under microwave heating

A mixture of 6-methoxyisatin (0.18g, 0.001 mol) and thiosemicarbazide (0.1g, 0.0011 mol) in presence of heterogeneous catalyst Mont. K-10, Mont. KSF was irradiated under microwave irradiation at 560W for 5-minutes. A yellow coloured solid formed during irradiation. The product formed is extracted using ethanol-DMF. Yield 0.252g (97% with Mont. K-10); 0.255g (98% with Mont. KSF), m.p. 265°C. [Found : N, 22.68, S, 12.62.  $C_{10}H_{10}N_4O_2S$  requires N, 22.40; S, 12.80%]; IR: 825, 860 (1, 2, 4-trisubstituted benzene ring), 1115 (C=S), 1125 & 1370 (C-O-C stretching), 1620 (C=N), 1700 (C=O), 3200, 3280, 3400 (NH, NH<sub>2</sub>).

### Acknowledgment

We thank Professor D. Villemin (France), Dr. R. Sharma (Dayton, USA) and Professor A.J. Bellamy (Swindon, UK) for inspiration.

# References

- [1] Mohan, J.; Anjaneyulu, G.S.R.; Kiran, Indian J. Chem. 1988, 27B, 128.
- [2] Snyder, H.R.; Benjamin, L.E., J. Med. Chem., 1966, 9, 402.
- [3] Pharmacal, N.; Neth. Appl. 1964, 6, 380; Chem. Abstr., 1965, 62,2780.
- [4] Trepanier, D.L.; Krieger, P.E., U.S. Pat., 1972, 3, 641; Chem. Abstr., 1972, 76, 127024k.
- [5] Sharpe, C.J.; Shadbolt, R.S.; Ashford, A.; Ross, J.W.; J. Med. Chem., 1971, 14, 977.
- [6] Kochhar, M.M.; Williams, B.B.; J. Med. Chem., 1972, 15, 322.
- [7] Kano, S.; Noguchi, T.; Japan Pat., 1971, 71, 836; Chem. Abstr., 1972, 76, 25295g.
- [8] Renfrew, E.E.; Pons, H.W.; U.S. Pat., 1976, 3, 130; Chem. Abstr., 1976, 85, 22753e.
- [9] Jenkins, P.W.; Brooker, L.G.S.; U.S. Pat., 1972, 3, 81; Chem. Abstr., 1973, 78, 85949z.
- [10] Hinata, M.; SHiba, K.; Takei, H.; Sato, A.; Sakai, T.; Ger. Offen., 1974, 2, 418; Chem. Abstr., 1975, 82, 49815K.
- [11] Brooker, L.G.S.; U.S. Pat., 1937, 2,729; Chem. Abstr., 1937, 31, 6989.
- [12] (a) Romagnoli, R; Baraldi, P.G; Cruz-Lopez, C; Preti, D; Bermejo, J; Estavez, F.; Chem. Med. Chem., 2009, 4, 1668.

(b) Bursavich, M.G.; Gilbert, A.M.; Lombardi, S; Georgiadis, K. E; Reifenberg, E, Flannery, C. R; Morris, E.A; *Bio-org. Med. Chem. Lett.* 2007, *17*, 5630.

(c) Konkel, M.J; Packiarajan, M; Chem. H Topiwala, U.P.; Jimenez, H; Talisman, I.J; Coate, H; Walker, M.W.; *Bioorg. Med. Chem. Lett*, 2006, *16*, 3950.

(d) Lam, P.Y.S; Vinoent, G; Clark, C.G; Dcudon, S; Jadhav, P.K.; Tetrahedron Lett.; 2001, 42, 3415.

- [13] (a) Shindikar, A.V; Khan, F; Viswanathan, C.L.; *Eur. J. Med. Chem.* 2006, *41*, 786.
  (b) Moser, P; Sallmann, A; Wieserberg, I.; *J. Med. Chem.* 1990, *33*, 2358.
  (c) Sarges, R; Howard, H. R; Koe. H.K; Weissman, A., *J. Med. Chem.* 1989, *32*, 437.
- [14] Peet, N.; J. Heterocyclic Chem.; 1990, 17, 1514.
- [15] For reviews, see: (a) Chem, Y; Larock, R.C.; In Modern Arylation Methods, Ackerman, J.; ED Wiley/ VCH New York, 2009, 401. (b) Sanz, R.;Org. Prep. Proced. Int.; 2008, 40, 215.
- [16] (a) Lin, Z; Larock, R. C.; J. Org. Chem. 2006, 71, 3198.
  (b) Lin, Z; Larock, R. C.; Org. Lett. 2003, 5, 4673.
  (c) Lin, Z; Larock, R. C.; Org. Lett. 2004, 6, 99.
- [17] Lin, Z; Larock, R. C.; J. American Chem. Soc.; 2005, 127, 13112.
- [18] Pintori, D. G; Greaney, M.F.; Org. Lett. 2010, 12, 168.
- [19] Yoshida, H; Shirakawa, E; Honda, Y; Hiyama, T.; Angew. Chem, Inted.; 2002, 41, 3246.
- [20] Zhao, J; Larock, R.C.; J.Org. Chem.; 2007, 72, 583.
- [21] Rogness D. C; Larock, R.C.; Tetrahedron Lett.; 2009, 50, 4003.
- [22] Lin, Z; Larock, R.C.; J. Org. Chem.; 2006, 71, 3198.
- [23] Peddibhotla, S.; Curr. Bioact. Compd.; 2009, 5, 20.

© 2015, by the Authors. The articles published from this journal are distributed to the public under "**Creative Commons Attribution License**" (http://creativecommons.org/licenses/by/3.0/). Therefore, upon proper citation of the original work, all the articles can be used without any restriction or can be distributed in any medium in any form.

#### Publication History

Received	$21^{st}$	Jan	2015
Revised	$20^{\text{th}}$	July	2015
Accepted	$12^{\text{th}}$	Aug	2015
Online	$30^{\text{th}}$	Aug	2015