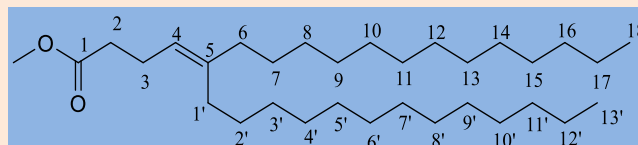


Research Article

Isolation and Structure Elucidation of Two New Compounds from Stem bark of *Prosopis cineraria*Lokesh Kumar Soni¹, Ashok Basak¹, Pradeep Parasher² and Mahabeer Prasad Dobhal^{1,*}¹Natural Products Laboratory, Department of Chemistry, University of Rajasthan²Department of Chemistry, Govt. P.G. College, Jhalawar**Abstract**

In the course of phytochemical study of chloroform extract of stem bark of *prosopis cineraria*, five compounds were isolated, (1) Methyl 5-tridecyloctadec-4-enoate, (2) nonacosan-8-one, (3) lupeol, (4) β -sitosterol and (5) stigmasterol. Among these, two compounds (1) Methyl 5-tridecyloctadec-4-enoate and (2) nonacosan-8-one have been isolated for the first time. Structure of these compounds was identified using spectral data analysis viz. IR, ¹H NMR, ¹³C NMR and Mass.

Keywords: *Prosopis cineraria*, Methyl 5-tridecyloctadec-4-enoate, nonacosan-8-one, lupeol, β -sitosterol, stigmasterol



Methyl 5-tridecyloctadec-4-enoate (1)

***Correspondence**

Author: Mahabeer Prasad Dobhal

Emails: mpdobhal@yahoo.com, lsoni492@gmail.com

Introduction

Prosopis cineraria (L.) Druce belongs to the cosmopolitan genus *Prosopis*, subfamily Mimosaceae, tribe Leguminosae/ Fabaceae [1]. The genus *Prosopis* includes 44 species of trees and shrubs [2]. *Prosopis cineraria* is five to ten meters in height and grows in dry and arid regions of Arabia and in some regions of Indian states, particularly Rajasthan, Haryana, Punjab, Gujarat, Western Uttar Pradesh and in dry areas of Deccan. It is known under numerous names, such as Janti and Chonksa (Delhi), Jhind, Jhand and Jand (Punjab and Haryana), Banni (Karnataka), Sumri (Gujarat), Kandi (Sindh) and Khejri (Sanskrit). The importance of the healthful worth of this tree has been highlighted in ancient Ayurvedic (medicine) literature. Its flower is pounded, mixed with sugar and used throughout maternity as safeguard against miscarriage. The bark of *Prosopis cineraria* is dry, acrid, bitter with a pointy taste; cooling anthelmintic, tonic, cures infectious disease, dysentery, bronchitis, asthma, leucoderma, piles, tremors of the muscles [3]. The bark is employed in rheumatism, cough and colds, diarrhea, worm infestations, and skin problems [4]. The bark of the plant offers immediate relief to an individual bitten by a snake or a scorpion [5].

According to literature survey various phytoconstituents have been isolated from different parts of *P. cineraria*. Some of them are patuletin glycoside patulitrin [6], sitosterol, spicigerine, flavone derivatives prosogerin A and prosogerin B [7], steroids like campesterol, cholesterol, sitosterol and stigmasterol, actacosanol, hentriacontane, methyl docosanoate, diisopropyl-10, 11-dihydroxyicosane-1,20-dioate, tricosan-1-ol, 7,24-tirucalladien-3-one along with a piperidine alkaloid spicigerine [8-9], prosogerin C [10], prosogerin D [11], prosogerin E, gallic acid, patuletin, patulitrin, luteolin, rutin [12-13], 3-benzyl-2-hydroxy-urs-12-en-28-oic acid, maslinic acid 3-glucoside, linoleic acid, prosphylline, 5,5'-oxybis-1,3-benzendiol, 3,4,5, trihydroxycinnamic acid 2-hydroxy ethyl ester, 5,30,40-trihydroxyflavanone 7-glycoside [14], methyl heptacosanoate, heneicosanoic acid, 4-hydroxy benzoic acid, methyl 4-hydroxycinnamate, methyl 2-methoxy-5-hydroxycinnamate and O-coumaroylglycerol [15].

Pharmacological activities like analgesic [16-18], antipyretic, anthelmintic [19], antioxidant [20], antimicrobial [21-23], anticonvulsant [24], antitumour [25,26], antihyperglycemic [27], antihyperlipidemic [28], antidiarrhoeal [29], hypolipidemic [30], antidepressant and skeletal muscle relaxant [31] and hepatoprotective activities [32] have been reported from different plant extracts of *P. cineraria*.

Materials and Methods

General experimental procedure

Column chromatography was performed on column (length 120 & diameter 2.5 cm), silica gel (60-120 mesh) and TLC on Merck's silica gel 60 F254 pre-coated glass plates. IR spectra were recorded on SHIMADZU FTIR-8400S spectrophotometer using KBr pellets. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectra (300 MHz and 75 MHz respectively) were recorded on Bruker spectrometer using CDCl_3 as solvent and TMS as internal standard. FAB mass spectra were recorded on JEOL SX 102/BA-600 mass spectrometer.

Extraction and isolation

The stem bark of the plant was collected from the locality of Bagru, Jaipur, India. The plant was identified by Mr. Vinod Kumar Sharma (voucher no. RUBL 211353) and a voucher specimens was deposited at the Herbarium of the Department of Botany, University of Rajasthan, Jaipur, India. Shade dried stem bark of *Prosopis cineraria* were powdered and extracted with methanol on a water bath for 48 hours. The extract was concentrated under reduced pressure to yield 300 g and then fractionated with petroleum ether (7 g), chloroform (20 g) and ethyl acetate (32 g). The chloroform soluble fraction (20 g) was chromatographed over silica gel (60-120 mesh) and eluted with the solvent, of increasing polarity. After elution compounds 1 to 3 were isolated and purified by PTLC followed by recrystallization. All the compounds were identified by $^1\text{H NMR}$, $^{13}\text{C NMR}$, and Mass spectral data (**Table 1**).

Table 1 ^1H , ^{13}C NMR data of compounds 1, 2 and 3

C- atom	1		2		3	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1	–	174.38	0.88 (<i>t</i> , <i>J</i> =6.9 Hz)	14.10	α , 1.68, <i>m</i> β , 1.20, <i>m</i>	27.43
2	2.32 (<i>t</i> , <i>J</i> =7.5 Hz)	34.14	1.03, <i>m</i>	22.68	α , 1.22, <i>m</i> β , 1.66, <i>m</i>	27.99
3	2.03 (<i>t</i> , <i>J</i> =7.5 Hz)	24.98	1.18, <i>m</i>	24.72	3.21, <i>m</i>	79.02
4	5.34 (<i>t</i> , <i>J</i> =5.4 Hz)	129.76	1.25, <i>m</i>	24.72	–	47.99
5	–	130.0	1.48, <i>m</i>	31.92	1.65, <i>m</i>	48.32
6	2.03 (<i>t</i> , <i>J</i> =5.7 Hz)	31.94	1.63, <i>m</i>	31.92	α , 1.13, <i>m</i> β , 1.63, <i>m</i>	31.92
7	1.38, <i>m</i>	29.71	2.34 (<i>t</i> , <i>J</i> =7.8 Hz)	42.62	α , 1.62, <i>m</i> β , 1.17, <i>m</i>	22.69
8	1.25, <i>m</i>	29.54	–	210.0	–	50.46
9	1.03, <i>m</i>	29.47	2.34 (<i>t</i> , <i>J</i> =7.8 Hz)	42.62	1.65, <i>m</i>	48.32
10	1.03, <i>m</i>	29.38	1.63, <i>m</i>	31.92	–	40.85
11	1.03, <i>m</i>	29.27	1.48, <i>m</i>	31.92	α , 1.63, <i>m</i> β , 1.07, <i>m</i>	25.16
12	1.03, <i>m</i>	29.27	1.25, <i>m</i>	29.07	α , 1.63, <i>m</i> β , 1.07, <i>m</i>	25.16
13	1.03, <i>m</i>	29.27	1.25, <i>m</i>	29.07	1.65, <i>m</i>	50.46
14	1.03, <i>m</i>	29.27	1.25, <i>m</i>	29.07	–	43.01
	1.03, <i>m</i>	29.17	1.25, <i>m</i>	29.07	α , 1.57, <i>m</i> β , 1.31, <i>m</i>	38.87
15					β , 1.31, <i>m</i>	
16	1.03, <i>m</i>	22.70	1.25, <i>m</i>	29.07	α , 1.57, <i>m</i> β , 1.31, <i>m</i>	25.16
17	1.03, <i>m</i>	22.70	1.25, <i>m</i>	29.07	–	42.84
18	0.88 (<i>t</i> , <i>J</i> =3.0 Hz)	14.13	1.25, <i>m</i>	29.07	1.61, <i>m</i>	55.31
19	–	–	1.25, <i>m</i>	29.07	2.37, <i>m</i>	55.31

20	–	–	1.25, <i>m</i>	29.07	–	150.97
21	–	–	1.25, <i>m</i>	29.07	α , 1.58, <i>m</i> β , 1.17, <i>m</i>	29.86
22	–	–	1.25, <i>m</i>	29.0	α , 1.58, <i>m</i> β , 1.17, <i>m</i>	31.92
23	–	–	1.25, <i>m</i>	29.07	1.03, <i>s</i>	27.43
24	–	–	1.25, <i>m</i>	29.07	0.96, <i>s</i>	14.46
25	–	–	1.25, <i>m</i>	29.07	0.94, <i>s</i>	17.90
26	–	–	1.25, <i>m</i>	29.07	0.82, <i>s</i>	16.42
27	–	–	1.18, <i>m</i>	29.07	0.78, <i>s</i>	16.10
28	–	–	1.03, <i>m</i>	22.67	0.72, <i>s</i>	19.25
29	–	–	0.88 (<i>t</i> , $J=6.9$ Hz)	14.1	2.10, <i>s</i>	21.70
30	–	–	–	–	<i>Ha</i> , 4.69, <i>brs</i> <i>Hb</i> , 4.57, <i>brs</i>	109.32
OCH_3	3.66, <i>s</i>	51.44	–	–	–	–
1'	2.03 (<i>t</i> , $J=5.7$ Hz)	31.94	–	–	–	–
2'	1.38, <i>m</i>	29.61	–	–	–	–
3'	1.25, <i>m</i>	29.54	–	–	–	–
4'	1.03, <i>m</i>	29.34	–	–	–	–
5'	1.03, <i>m</i>	29.34	–	–	–	–
6'	1.03, <i>m</i>	29.34	–	–	–	–
7'	1.03, <i>m</i>	29.27	–	–	–	–
8'	1.03, <i>m</i>	29.27	–	–	–	–
9'	1.03, <i>m</i>	29.17	–	–	–	–
10'	1.03, <i>m</i>	22.70	–	–	–	–
11'	1.03, <i>m</i>	22.70	–	–	–	–
12'	1.03, <i>m</i>	22.70	–	–	–	–
13'	0.88 (<i>t</i> , $J=3.0$ Hz)	14.13	–	–	–	–

Isolation of Methyl 5-tridecyloctadec-4-enoate (1)

On eluting the column with petroleum ether-chloroform (1:1) yielded Methyl 5-tridecyloctadec-4-enoate and recrystallization from acetone gave sticky off-white compound. It showed homogenous behaviour on TLC. The spectral data are as: IR (KBr) 1735, 1300, 810 cm^{-1} . API- Mass (m/z) 478.2 [M^+], 423.3, 395.3, 281.2, 255.3 (base peak).

Isolation of Nonacosan-8- one (2)

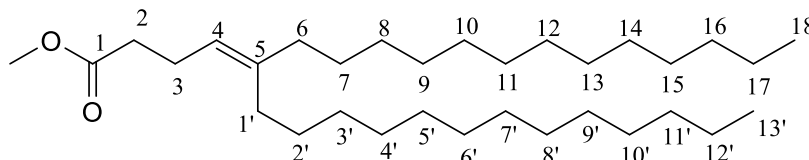
Elution of column with petroleum ether- chloroform (1:4) yielded Nonacosan-8- one and recrystallized purified from acetone. It is a white powderd compound and showed homogenous behaviour on TLC. The spectral data are obtained as: IR (KBr) 1720, 2950 cm^{-1} . API-Mass (m/z) 423.3 (M^+), 395.3, 339.3, 325.0 (base peak), 311.0, 297.0, 255.3.

Isolation of Lupeol (3)

Further elution of column with petroleum ether- chloroform (2:3) yielded the triterpenoid lupeol after purification through PTLTLC. The spectral data are as: IR (KBr) 3450, 2950, 2850, 1640, 1385, 1360, 1310, 1290, 1270, 1110, 880. API-Mass (m/z) 426.0 (M^+), 408.4, 393.4, 327.2, 229.2, 218.1, 204.1, 189.1 (base peak), 175.1, 161.1, 147.1, 121.1, 107.0.

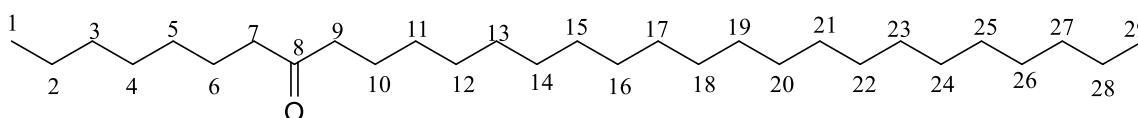
Results and Discussion

Compound (1) was obtained as sticky pale white. Its mass spectrum showed $[M^+]$ at 478.2 corresponding to molecular formula $C_{32}H_{62}O_2$. IR spectrum (KBr) of compound **1** showed the presence of ester group by showing absorption at 1735 cm^{-1} . It showed characteristic absorptions at $2900, 2840\text{ cm}^{-1}$ for C-H stretching and 1250 cm^{-1} for C-O stretching of the ester group. The weakly intense absorption observed at 1590 cm^{-1} indicate the presence of C=C bond. The $^1\text{H-NMR}$ spectrum shows one singlet appeared at $\delta 3.66$ assigned to methyl group of ester oxygen (R-COO-CH₃). Presence of a triplet at $\delta 5.34$ ($J = 5.4\text{ Hz}$, 1H, C-4) confirming the presence of a double bond. Two triplets were observed at $\delta 2.32$ ($J = 7.5\text{ Hz}$) and $\delta 2.03$ ($J = 5.7\text{ Hz}$) were assigned to a methylene protons (C-2) attached to the ester linkage and methylene protons (C-3) attached to double bond respectively. Two singlets appeared at $\delta 0.85$ and $\delta 0.88$ were assigned to C-13' and C-18 methyl group protons respectively. In $^{13}\text{C-NMR}$ spectrum of compound **1** the carbonyl carbon of ester appeared at $\delta 174.38$ and it was assigned to C-1 position. The double bonded carbons appeared at $\delta 129.7$ and $\delta 130.0$, and assigned to C-4 and C-5 carbon atoms respectively.



Methyl 5-tridecyloctadec-4-enoate (1)

Compound (2) was obtained as a white powder. The molecular formula was determined as $C_{29}H_{58}O$ by API-ES mass spectroscopy $[M^+]$ m/z 423.3. The other prominent ions were observed at m/z 395.3, 339.3, 325.0 (base peak), 311.0 etc. $^1\text{H NMR}$ spectrum showed the presence of two methyl groups by showing the singlets at $\delta 0.88$ (t, C-1) and 0.85 (t, C-29) for three protons each. Two triplets of equal intensity and coupling constant were observed at $\delta 2.34$ and assigned to two methylene protons attached to the carbonyl linkage (-CH₂-CO-CH₂-). A broad singlet at $\delta 1.24$ showed the presence of remaining 48 H for 24 (CH₂) groups. In ^{13}C NMR spectrum absorbance at $\delta 210$ indicate the presence of carbonyl carbon and assigned at C-8 position. IR spectrum (KBr) of compound (2) was confirmed the presence of carbonyl group by showing the absorbance 1720 cm^{-1} . It indicates characteristic absorbance at $2900, 2850$ for C-H stretching.

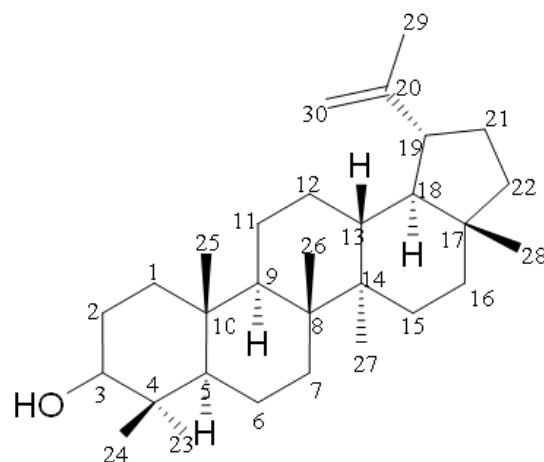


nonacosan-8-one (2)

Compound (3) was obtained as white niddles. It's molecular formula was determined by API-Mass spectrum as $C_{30}H_{50}O$, showing M^+ at m/z 426.0. The other important peaks were observed at m/z 408.46 $[M^+ - H_2O]$, 229.23 $[C_{17}H_{25}^+]$, 189.15 $[C_{14}H^+_{21}]$, base peak] indicating the presence of a lupane skeleton with a hydroxyl group in ring A [33].

In IR spectrum (KBr) of compound (3) broad absorption was appeared at 3450 Cm^{-1} indicating the presence of O-H stretching. Absorption at 1640 and 880 Cm^{-1} is related to C=CH₂ stretching. Absorptions at $1385, 1360, 1310, 1290$ and 1270 are characteristic of lupine series [34]. Absorption at 1110 cm^{-1} indicates the presence of C-O stretching.

$^1\text{H NMR}$ spectrum showed the presence of six methyl groups as singlets at $\delta 1.03, 0.96, 0.94, 0.82, 0.78$ and 0.72 while the olefinic methyl group displayed a singlet at $\delta 2.1$. A pair of downfield broad singlet appeared at $\delta 4.69$ and $\delta 4.57$ showed the presence of vinylic protons. A multiplet was appeared at $\delta 3.21$ is due to H-3 proton. The multiplet in the region $\delta 1.20-1.68$ showed the presence of remaining 24 protons. In ^{13}C NMR spectrum the signal due to an exomethylene group appeared at $\delta 109.32$ and $\delta 150.97$ and assigned at C-20 and C-30 position respectively. The absorbance for seven methyl groups appeared at $\delta 27.3$ (C-23), $\delta 14.46$ (C-24), $\delta 17.9$ (C-25), $\delta 16.4$ (C-26), $\delta 16.1$ (C-27), $\delta 19.25$ (C-28), $\delta 19.5$ (C-29). The deshielded signal appearing at 79.02 showed the presence of hydroxyl group at C-3 position.



Lupeol (3)

Acknowledgement

The author (LKS) is grateful to CSIR, New Delhi for the senior research fellowship.

References

- [1] Rasanen LA, Lindstrom K, Ind J Exp Biol 2003, 41(10), 1142-59.
- [2] Ramirez L, Vega ADL, Razkin N, Luna V, Harris PJC, Agronomie 1999, 19, 31-43.
- [3] Kirtikar KR, Basu BD, Indian Medicinal Plants 1984, 2, 910-911.
- [4] Sharma AK (Ed.), Diabetes mellitus and its complications: An update, 1ed. Sharma AK, Macmillan India Ltd, New Delhi, 1993, p 92-205.
- [5] Chopra RN, Nayar SL, Chopra IC, Glossary of Indian Medicinal Plants, CSIR, New delhi, 1956, p 204.
- [6] Nandkarni KM, Ind Mat Med, Popular prakashan, Mumbai, Vol. 1, 2000, p. 101.
- [7] Sharma RC, Zaman A, Kidwai AR, Chemical examination of Prosopis spicigera Linn. Indian J Chem 1964, 2(2), 83-84.
- [8] Bhardwaj DK, Bisht MS, Mehta CK, Sharma GC, Phytochemistry 1979, 18, 355-366.
- [9] Malik A, Kalidhar SB, Indian J Pharm Sci 2007, 69(4), 576-578.
- [10] Jewers K, Nagler MJ, Zirvi KA, Amir F, Phytochemistry 1976, 15, 238-240.
- [11] Bhardwaj DK, Jain RK, Sharma GC, Mehta CK, Indian J Chem Sec B 1978, 16, 1133-1134.
- [12] Bhardwaj DK, Bisht MS, Jain RK, Phytochemistry 1980, 19, 1269-1270.
- [13] Bhardwaj DK, Gupta AK, Jain RK, Sharma GC, J Nat Prod 1981, 44(6), 656-659.
- [14] Ukani MD, Limbani NB, Mehta NK, Anc Sci of Life 2000, 20(1), 1-13.
- [15] Liu Y, Singh D, Nair MG, J Funct Foods 2012, 4, 116-121.
- [16] Khan ST, Riaz N, Afza N, Nelofar A, Malik A, Ahmed E, J Chem Soc of Pak 2006, 28(6), 619-622.
- [17] Kumar A, Yadav SK, Singh S, Pandeya SN, J Appl Pharm Sci 2011, 1, 158-160.
- [18] Muzammil A, Farhana T, Salman A, Int Res J Pharm 2013, 4, 93.
- [19] Ramasamy VMM, Venugopalan R, Ramnathan SK, Perumal P, Chellapan DR, J Pharm Res 2009, 2, 660-662.
- [20] Velmurugan V, Arunachalam G, Ravichandran V, Asian J Pharm Sci Res 2011, 1, 88-91.
- [21] Dharani B, Sumathi S, Sivaprabha J, Padma PR, J Nat Prod Plant Resour 2011, 1, 26-32.
- [22] Sharma R, Jodhawat N, Purohit S, Kaur S, Int J Pharm Sci Rev Res 2012, 14, 15-17.
- [23] Velmurugan V, Arunachalam G, Ravichandran V, Arch Appl Sci Res 2010, 2, 147-150.
- [24] Aneela S, Dey A, Babu AMMS, De S, Int J Chem Pharm Sci 2014, 5, 42.
- [25] Velmurugan V, Arunachalam G, Ravichandran V, Int J Pharm Res 2012, 4, 88-91.
- [26] Robertson S, Narayanan N, Kapoor BR, Nat Prod Res 2011, 25, 857-862.
- [27] Maideen NMP, Velayutham R, Manavalan G, Asian J Pharm Life Sci 2012, 2, 1-9.

- [28] Sharma D, Singla PY, J Sci Inn Res 2013, 2, 751.
[29] Naik ND, Malothu R, Reddy RG, Naadella BC, Jayasri P, Elumalai A, Int J Bio Pharm Res 2012, 3, 317.
[30] Purohit A, Ram H, Asian J Pharm Clin Res 2012, 5, 106-109.
[31] George M, Joseph L, Sharma A, Brazilian J Pharm Sci 2012, 48, 577-581.
[32] Velmurugan V, Arunachalam G, Int J Pharm Sci 2014, 6, 491-493.
[33] Shiojima K, Arai Y, Masudak K, Takase Y, Ageta T, Ageta H, Chem Pharm Bull 1992, 40, 1683-1690.
[34] Snatzke G, Lampert F, Tschesche R, Tetrahedron, 1962, 18, 1417-1431.

© 2015, by the Authors. The articles published from this journal are distributed to the public under “**Creative Commons Attribution License**” (<http://creativecommons.org/licenses/by/3.0/>). Therefore, upon proper citation of the original work, all the articles can be used without any restriction or can be distributed in any medium in any form.

Publication History

Received 03rd June 2015
Revised 18th June 2015
Accepted 16th July 2015
Online 30th July 2015